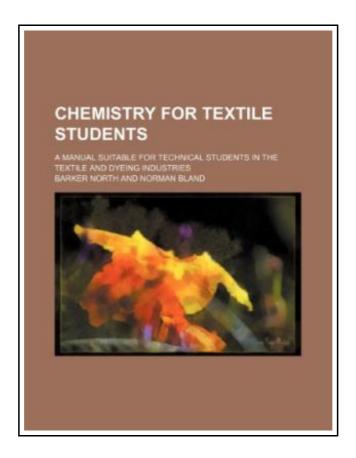
Chemistry for Textile Students A Manual Suitable for Technical Students in the Textile and Dyeing Industries



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Reviews

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(Horace Schroeder)

CHEMISTRY FOR TEXTILE STUDENTS A MANUAL SUITABLE FOR TECHNICAL STUDENTS IN THE TEXTILE AND DYEING INDUSTRIES



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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 134 pages. Dimensions: 9.6in. x 7.4in. x 0.6in.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1920 Excerpt: . . . quite brittle and opaque. (4) 8 Sulphur or milk of sulphur. S sulphur is obtained as a white powder when hydrochloric acid acts on a solution containing calcium polysulphides, or when sulphur is precipitated by passing sulphuretted hydrogen into an oxidising agent. The properties of the allotropic modifications of sulphur are shown in the following table: Sulphur is used in the manufacture of matches, gunpowder, dyestuffs, for the production of sulphur dioxide for bleaching wool and for the manufacture of sulphuric acid. All the forms of sulphur burn in oxygen or air, with the production of sulphur dioxide. Milk of sulphur is used in pharmacy. Sulphuretted hydrogen, hydrogen sulphide or hydrosulphuric acid, H2S. Preparation. Sulphuretted hydrogen is formed in small quantities when a stream of hydrogen is passed through boiling sulphur. Pure sulphuretted hydrogen is made by gently heating strong hydrochloric acid with antimony sulphide. Sb2S3 6HCl 2SbCl3 3H2S It is nearly always prepared, however, by the action of dilute sulphuric or hydrochloric acid on ferrous sulphide, an apparatus similar to that used for the preparation of hydrogen being employed. FeS 2HCl-. FeCl2 H2S Sulphuretted hydrogen prepared in this way is always impure, containing hydrogen, and traces of other gases, due to the presence of iron in small quantities in commercial ferrous sulphide. Properties. Sulphuretted hydrogen is a colourless gas, with a sweetish taste and a very offensive odour. It can be liquefied at-62 C. It is soluble in water, one volume of water at 15...

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